# AGRICULTURAL AND FOOD CHEMISTRY

# Effect of High Oxygen Atmosphere Storage on Quality, Antioxidant Enzymes, and DPPH-Radical Scavenging Activity of Chinese Bayberry Fruit

Zhenfeng Yang,<sup> $\dagger,\ddagger$ </sup> Yonghua Zheng,<sup> $*,\dagger$ </sup> and Shifeng Cao<sup> $\dagger$ </sup>

College of Food Science and Technology, Nanjing Agricultural University, Nanjing 210095, the People's Republic of China, and College of Biological and Environmental Sciences, Zhejiang Wanli University, Ningbo 315100, the People's Republic of China

The influence of high  $O_2$  atmosphere on postharvest decay, quality, total phenolic, total anthocyanin contents, antioxidant enzymes activity, and antioxidant activity of Chinese bayberry fruit was investigated. Freshly harvested Chinese bayberry fruits were placed in jars and ventilated continuously with air or with 80 and 100%  $O_2$  for up to 12 days. Samples were randomly selected initially and at 3-days interval during storage. The fruit exposed to high  $O_2$  was resistant to decay, had high levels of total soluble solids, titratable acidity and ascorbic acid contents, and also reduced the increment of pH value. High  $O_2$  treatment was less stressful as reflected by having the significantly lower malonaldehyde contents and higher catalase, ascorbic acid peroxidase, and peroxidase activities during storage. Both 80% and 100%  $O_2$  treatments had also retained the bioactive contents and 1,1-diphenyl-2-picrylhydrazyl radical scavenging activity during storage. These results indicate that elevated  $O_2$  levels may improve the ability of the antioxidative defense mechanism in Chinese bayberry and result in a better control of fruit decay.

### KEYWORDS: Chinese bayberry; high O2 atmosphere; antioxidant enzymes; antioxidant activity; fruit decay

## INTRODUCTION

Chinese bayberry (*Myrica rubra* Sieb. & Zucc.) is a tropical or subtropical native fruit in China with high commercial value for its red to purple color and appealing taste. To obtain the best flavor, Chinese bayberry is commercially harvested as almost fully ripe from mid June to early July. However, at this stage, the fruit is soft and highly perishable, susceptible to mechanical injury, physiological deterioration, water loss, and microbiological decay, with limiting postharvest life to 1-2 days under ambient temperature, which has resulted in a reduced market value (1).

Storage in lower temperature combined with conventional low  $O_2$  controlled atmosphere storage has been reported to reduce Chinese bayberry fruit decay (2). However, the growth of anaerobic fungi and accumulation of anaerobic fermentation products can influence its storage life and flavor (3). More recently, elevated  $O_2$  modified atmosphere has been shown to prolong the shelf life of various horticultural products (4–8). It was suggested that high  $O_2$  treatment resulted in the suppression of microbial growth and therefore retarded decay in fruit (8). However, the mechanisms by which high  $O_2$ atmosphere inhibit fruit decay are yet unclear.

Elevated O<sub>2</sub> concentrations can cause the production of reactive oxygen species (ROS), such as superoxide, hydrogen peroxide, and the hydroxyl radical, thus damaging plant tissues (9). The sensitivity to  $O_2$  toxicity varies among plant species (4). Defense against oxidative stress in plants to prevent or alleviate the damage from ROS includes enzymatic ROS scavenging systems and nonenzymatic antioxidant compounds. The enzymatic ROS scavenging systems included superoxide dismutase (SOD) and catalase (CAT) (10, 11), the glutathione peroxidase system and the ascorbate-glutathione cycle (12, 13). The behavior of water and lipid soluble antioxidants, such as ascorbic acid, glutathione,  $\alpha$ -tocopherol, carotenoids, and various types of secondary metabolites, mostly composed of total phenolics compounds such as flavones, flavonols, and anthocyanins have also been linked to function as ROS scavengers. In previous work, high  $O_2$  atmosphere storage effectively inhibited the decay of blueberries and strawberries (7, 14). But it is still unknown how high O2 ameliorates fruit decay, and there were no reports published about the effects of high O<sub>2</sub> atmosphere storage on antioxidant enzymes and nonenzymatic antioxidants in Chinese bayberry.

The objective of this study was to investigate the effect of high  $O_2$  treatment on antioxidant status of Chinese bayberry in association with fruit decay and fruit quality during storage and then to understand the role of high  $O_2$  atmosphere in bayberry fruit decay control.

10.1021/jf803007j CCC: \$40.75 © 2009 American Chemical Society Published on Web 12/18/2008

<sup>\*</sup> To whom correspondence should be addressed. Tel: (86) 25-884399080. E-mail: zhengyh@njau.edu.cn.

<sup>&</sup>lt;sup>†</sup> Nanjing Agricultural University.

<sup>&</sup>lt;sup>‡</sup>Zhejiang Wanli University.

#### MATERIALS AND METHODS

**Plant Material and Treatment.** Chinese bayberry (*Myrica rubra* Sieb. & Zucc. cv. Wumei) was hand-harvested from a commercial orchard in Suzhou (June 2006), Jiangsu province, and transported to the laboratory within 4 h. Fruit was sorted to eliminate damaged, unripe fruit and selected for uniform size and color. Two kilograms of fruit were placed in a 8-L jar, and three jars were used for each treatment. The jars were placed at 5 °C and connected to a continuous flow (40 mL min<sup>-1</sup>) of humidified air (control), 80% and 100% O<sub>2</sub> (balance N<sub>2</sub> in all high O<sub>2</sub> treatments). Oxygen concentration was checked regularly with an O<sub>2</sub>/CO<sub>2</sub> analyzer (AMETEK, Pittsburgh, PA) and maintained at  $\pm$  2% during storage. Samples were taken at 3-day intervals during storage for decay evaluation. At each time point, pulp tissues were taken from each fruit, frozen in liquid nitrogen, and stored at -80 °C until analyzed.

**Fruit Decay.** The symptom of decay in Chinese bayberry during storage is visible mold growth on the fruit surface. Fruit decay was visually estimated by measuring the extent of decaying area on 15 fruits from each replicate and was determined by rating on a scale from 0 to 3, with 0 = normal (not decayed); 1 = slight (up to 15% of the surface affected); 2 = moderate (15–50% of the surface affected); 3 = severe (>50% of the surface affected). The fruit decay index was calculated using the following formula:  $[(1 \times N_1 + 2 \times N_2 + 3 \times N_3) \times 100/(3 \times N)]$ , where *N* is the total number of fruit measured, and  $N_1$ ,  $N_2$ , and  $N_3$  are the numbers of fruit showing different degrees of decay.

**Total Soluble Solids, Titratable Acidity, and pH.** Ten fruits from each replicate were wrapped in cheesecloth and squeezed with a hand press, and the juice was analyzed for total soluble solids (TSS), titratable acidity (TA), and pH. TSS was determined at 25 °C using a portable refractometer (WYT-4, Quanzhou, China). TA was determined by titrating 20 mL of bayberry juice to pH 8.2 using 0.2 mol  $L^{-1}$  NaOH, and pH was measured with a pH-meter (PHS-25B, Shanghai, China).

**Respiratory Rate and Ethylene Production.** Ten fruits for each of three replicates at each time point were enclosed in 250 mL glass jars at 5 °C, 10 mL of headspace gas was taken from each jar at the end of 2 h enclosure.  $CO_2$  was measured with an infrared gas analyzer (GXH-305, Beijing, China). Ethylene production was determined by gas chromatography using a flame ionization detector (Shimadzu 14B, Kyoto, Japan).

**Malondialdehyde (MDA) and Ascorbic Acid Content.** To analyze MDA content, 2 g of fresh tissue was homogenized with 5 mL of 5% (v/v) trichloroacetic acid (TCA) and then centrifuged at 12,000g for 10 min (4 °C). The MDA content was determined by adding 2 mL of 0.5% 2-thiobarbituric acid in 15% trichloroacetic acid to 1 mL of extraction. The mixture was heated at 95 °C for 20 min, quickly cooled in an ice-bath for 5 min, and then centrifuged at 12,000g for 10 min to clarify the solution. Absorbance at 532 nm was measured and subtracted from the absorbance at 600 nm. The amount of MDA was calculated with an absorptivity coefficient of 155 mmol cm<sup>-1</sup>.

Ascorbic acid was quantitatively determined by using 2,6-the dichlorophenolindophenol dye method as described by Jones and Hughes (15) with modifications. Fresh samples (10 g) were homogenized with 10 mL of 3% metaphosphoric acid (v/v). The extract was made up to a volume of 100 mL and centrifuged at 3000g for 15 min at room temperature. After decolorizing with 10 g of diatomite, 10 mL of supernatant was titrated against the standard 2,6-dichlorophenolindophenol dye.

**Enzyme Activities.** Two grams of pulp tissues was ground with a chilled mortar and pestle in 10 mL of pH 7.0, 50 mmol  $L^{-1}$  phosphate buffer or pH 7.5, Tris-HCl buffer (containing 3 mmol  $L^{-1}$  MgCl<sub>2</sub>, 0.1 mmol  $L^{-1}$  EDTA, and 1% insoluble polyvinylpyrrolidone) at 4 °C. The homogenate was then centrifuged at 20,000*g* for 20 min (4 °C), and the supernatant was collected for the enzyme activity assay. Protein was measured according to the method of Bradford (*16*), using bovine serum albumin (BSA) as the standard.

SOD activity was determined by the method of Rao et al. (17). The reaction medium contained 50 mmol L<sup>-1</sup> sodium phosphate buffer (pH 7.8), 14 mmol L<sup>-1</sup> methionine, 3  $\mu$ mol L<sup>-1</sup> EDTA, 1  $\mu$ mol L<sup>-1</sup> nitroblue-tetrazolium (NBT), 60  $\mu$ mol L<sup>-1</sup> riboflavin, and 100  $\mu$ L of enzyme extract. Three milliliters of the assay mixture in uniform, transparent

tubes was shaken and placed 50 cm below a light-bank consisting of four 30-W fluorescent lamps. The reaction was started by switching on the light; after 10 min, the light was turned off, and the absorbance by the assay mixture at 560 nm was recorded. One unit of SOD activity is defined as the amount of enzyme that inhibits the NBT photoreduction by 50% under the conditions of the assay.

CAT activity was determined according to Beers and Sizer (18) by monitoring the disappearance of  $H_2O_2$  by recording the decrease in absorbance at 240 nm of a reaction mixture containing 50 mmol L<sup>-1</sup> sodium phosphate buffer (pH 7.0), 12.5 mmol L<sup>-1</sup> H<sub>2</sub>O<sub>2</sub>, and 20  $\mu$ L of enzyme extract. One unit of CAT activity is defined as the amount of enzyme that decomposes 1  $\mu$ mol of H<sub>2</sub>O<sub>2</sub> per minute per milligram of protein under the conditions of the assay.

APX activity was measured as described by Nakano and Asada (19). The assay mixture consisted of 2.8 mL of 50 mmol L<sup>-1</sup> sodium phosphate buffer (pH 7.0), 100  $\mu$ L of 9 mmol L<sup>-1</sup> ascorbic acid, 10  $\mu$ L of 30% H<sub>2</sub>O<sub>2</sub>, and 100  $\mu$ L of enzyme extract. One unit of APX activity is defined as the amount of enzyme that oxidized 1  $\mu$ mol of ascorbate per minute per milligram of protein.

POD was assayed using the method of Kochba et al. (20) with modifications. The assay mixture (2 mL) consisted of 50 mmol  $L^{-1}$  sodium phosphate buffer (pH 6.5), 6 mmol  $L^{-1}$  guaiacol, 4.5 mmol  $L^{-1}$  H<sub>2</sub>O<sub>2</sub>, and 1 mL of crude enzyme extract. Increment in absorbance at 470 nm at intervals of 30 s was recorded spectrophotometrically. One unit of POD activity is defined as the amount of enzyme that catalyzes the peroxidation of 1 mmol of guaiacol per minute per milligram of protein.

Total Anthocyanin, Total Phenolic, and DPPH Radical Scavenging Activity. To prepare the fruit extract, 5 g samples from each replica were homogenized with 5 mL of precooled 95% ethanol containing 3% formic acid (v/v), and after centrifugation at 10,000g for 15 min (4 °C), another 15 mL of 80% ethanol containing 5% formic acid (v/ v) was used to extract the residue again. The supernatant was combined to make the final volume of 25 mL for analysis.

Total anthocyanin content of bayberry extract was measured using the pH differential method (21). Absorbance was measured at 510 and 700 nm, respectively, in different buffers at pH 1.0 and 4.5, using  $A = [(A_{510} - A_{700})_{\text{pH}.0} - (A_{510} - A_{700})_{\text{pH}4.5}]$  with a molar extinction coefficient of cyaniding 3-glucoside of 29600. Results were expressed as milligrams of cyaniding 3-glucoside (C 3-G) equivalents per gram of fresh weight.

Total soluble phenolic content in bayberry extract was determined according to the Folin–Ciocalteu procedure (22). To 0.2 mL of diluted extract, 1 mL of Folin–Ciocalteu reagent and 0.8 mL of  $Na_2CO_3$  (75 g L<sup>-1</sup>) were added. The mixture was incubated at 30 °C for 60 min. For the control, 0.2 mL of ethanol was used, and the absorbance was measured at 765 nm. The results were expressed as milligrams of gallic acid equivalent (GAE) per gram of fresh weight.

The antiradical capacity of the sample extract was estimated according to the 1,1-diphenyl-2-picrylhydrazyl (DPPH) assay (23). Briefly, an aliquot (0.1 mL) of the ethanol extract was added to 2.9 mL of DPPH (120  $\mu$ mol L<sup>-1</sup>) in methanol. A spectrophotometer (UV-754, Shanghai) was used, and the absorbance at 517 nm was measured after the reaction mixtures were incubated for 30 min at 30 °C in dark. Inhibition percentage was calculated using the equation: % inhibition = [(C - S)/C] × 100, where C is the net absorbance of the control, and S is the net absorbance of the sample.  $\alpha$ -tocopherol was used as a standard antioxidant analyzed at the same time. The final results were calculated and expressed as  $\alpha$ -tocopherol equivalents (TE) per gram on a fresh weight basis.

**Data Analysis.** Experiments were performed using a completely randomized design. All statistical analyses were performed using SPSS 13.0 (SPSS Inc., Chicago, IL). The data were analyzed by one-way analysis of variance (ANOVA). The treatment means were separated using Tukey's test, and differences at  $P \le 0.05$  were considered to be significant.

### **RESULTS AND DISCUSSION**

**Fruit Decay.** Fruit decay was markedly affected by different storage atmospheres (**Table 1**). Chinese bayberry stored under

Table 1. Changes in Chinese Bayberry Fruit Decay, Total Soluble Solids (TSS), Titratable Acidity (TA), and pH during Storage at 5 °C in Air or High  $O_2$  Atmosphere<sup>*a*</sup>

treatment	decay index (%)	TSS (%)	TA (%)	pН
day 0, air	0.00	$9.96\pm0.06$	$0.77\pm0.01$	$3.53\pm0.01$
day 3, air 80% O <sub>2</sub> 100% O <sub>2</sub>	$2.22 \pm 0.85~\text{a}$ 1.11 $\pm$ 0.90 a 0.00 a	$\begin{array}{c} 9.33 \pm 0.06 \text{ b} \\ 9.63 \pm 0.06 \text{ a} \\ 9.67 \pm 0.06 \text{ a} \end{array}$	$\begin{array}{c} 0.71 \pm 0.01 \text{ b} \\ 0.74 \pm 0.01 \text{ a} \\ 0.76 \pm 0.01 \text{ a} \end{array}$	$\begin{array}{c} 3.56 \pm 0.01 \text{ a} \\ 3.54 \pm 0.01 \text{ a} \\ 3.53 \pm 0.02 \text{ a} \end{array}$
day 6, air 80% O <sub>2</sub> 100% O <sub>2</sub>	$\begin{array}{c} 10.00 \pm 0.73 \text{ a} \\ 4.45 \pm 0.92 \text{ b} \\ 2.22 \pm 0.92 \text{ b} \end{array}$	$\begin{array}{c} 8.97 \pm 0.06 \text{ c} \\ 9.17 \pm 0.06 \text{ b} \\ 9.37 \pm 0.09 \text{ a} \end{array}$	$\begin{array}{c} 0.67 \pm 0.01 \text{ b} \\ 0.71 \pm 0.01 \text{ a} \\ 0.72 \pm 0.01 \text{ a} \end{array}$	$\begin{array}{c} 3.57 \pm 0.01 \text{ a} \\ 3.55 \pm 0.01 \text{ a} \\ 3.53 \pm 0.01 \text{ a} \end{array}$
day 9, air 80% O <sub>2</sub> 100% O <sub>2</sub>	$\begin{array}{c} \text{21.11} \pm \text{1.09 a} \\ \text{8.89} \pm \text{1.92 b} \\ \text{7.78} \pm \text{1.92 b} \end{array}$	$\begin{array}{c} 8.67 \pm 0.06 \text{ c} \\ 8.97 \pm 0.06 \text{ b} \\ 9.21 \pm 0.01 \text{ a} \end{array}$	$\begin{array}{c} 0.66 \pm 0.01 \text{ a} \\ 0.68 \pm 0.01 \text{ a} \\ 0.69 \pm 0.01 \text{ a} \end{array}$	$\begin{array}{c} 3.59 \pm 0.01 \text{ a} \\ 3.56 \pm 0.02 \text{ a} \\ 3.54 \pm 0.01 \text{ a} \end{array}$
day 12, air 80% O <sub>2</sub> 100% O <sub>2</sub>	$54.55 \pm 2.20$ a 22.22 $\pm$ 2.68 b 16.67 $\pm$ 1.23 c	$\begin{array}{c} 8.34 \pm 0.07 \text{ c} \\ 8.94 \pm 0.03 \text{ b} \\ 9.13 \pm 0.06 \text{ a} \end{array}$	$\begin{array}{c} 0.60 \pm 0.01 \text{ b} \\ 0.62 \pm 0.01 \text{ b} \\ 0.65 \pm 0.01 \text{ a} \end{array}$	$\begin{array}{c} 3.60 \pm 0.01 \text{ a} \\ 3.56 \pm 0.01 \text{ b} \\ 3.54 \pm 0.01 \text{ b} \end{array}$

 $^a$  Data were expressed as the mean  $\pm$  SEM of triplicate assays. Values in the same column with different letters for each day were significantly different at  $P \leq 0.05$ .

air showed slight fungal decay on day 3 but 55% decay on day 12 during storage at 5 °C. Fruit exposed to 80% and 100% O<sub>2</sub> exhibited visible fungal decay on day 6, and the decay rate increased gradually thereafter. After 12 days, the decay index of Chinese bayberry exposed to 80% and 100% O<sub>2</sub> was only 20% and 17%, respectively. No significant difference was noted in fruit decay between 80% and 100% O<sub>2</sub> treatments during storage. Similar results were obtained on strawberry (6) and blueberry (7). High  $O_2$  treatments (80–100%) were more effective in suppressing strawberry fruit decay caused by Botritis cinerea infection than in inhibiting the growth in vitro of Botritis cinerea (7), suggesting that the high oxygen atmospheres have an effect on the fruit itself as well as the decay caused by fungus, thereby resulting in greater effect in vivo than in vitro. However, the mechanism by which high O2 atmosphere inhibits fruit decay is still unclear.

TSS, TA, and PH. TSS contents of Chinese bayberry in all treatments decreased with the storage time (**Table 1**). High  $O_2$ atmosphere maintained higher TSS contents in comparison with that with air treatment. In blueberries, after storage of 28 days or longer, significantly higher values of TSS were maintained in fruit held at O<sub>2</sub> levels above 60% than fruit held at 40% and lower  $O_2$  levels (7). In contrast, significantly lower TSS values in high O<sub>2</sub> treated strawberries than in air-stored fruit during the later period of storage at 5 °C were reported in earlier studies (24). TA contents of Chinese bayberry decreased gradually during storage corresponding to the pH increase in all treatments. Fruits stored in high O<sub>2</sub> atmosphere tended to have higher TA content and lower pH values than control fruit. No significant differences were observed among all of the treatments on all sampling days. Similar results were also observed by Zheng et al. (7) in blueberries. However, Pérez and Sanz (24) found significantly higher TA content before day 4 and lower TA content after day 7 in strawberry fruit exposed to 90%  $O_2$  + 10% CO<sub>2</sub> than in fruit held in air during 9 days of storage at 8 °C. As the main substrates of respiratory metabolism, sugars and acids are depleted, causing corresponding changes in TSS, TA, and pH during storage. It has been shown that exposure of harvested horticultural crops to superatmospheric O<sub>2</sub> levels may stimulate, have no effect on, or reduce rates of respiration, depending on the commodity, maturity stage, time, and tem-

**Table 2.** Changes in Chinese Bayberry Fruit Respiratory Rate, EthyleneProduction, Ascorbic Acid, and Malondialdehyde (MDA) Content duringStorage at 5 °C in Air or High  $O_2$  Atmosphere<sup>a</sup>

treatment	respiratory rate (mg CO <sub>2</sub> kg <sup>-1</sup> FW h <sup>-1</sup> )	ethylene production ( <i>u</i> L kg <sup>-1</sup> FW h <sup>-1</sup> )	ascorbic acid (mg 100 g <sup>-1</sup> FW)	MDA (nmol g <sup>-1</sup> FW)
day 0, air	$\textbf{34.98} \pm \textbf{1.27}$	$23.67 \pm 0.56$	$65.53 \pm 1.30$	$1.64\pm0.55$
day 3, air 80% O <sub>2</sub> 100% O <sub>2</sub>	$\begin{array}{c} 26.60 \pm 1.01 \text{ a} \\ 24.70 \pm 0.77 \text{ b} \\ 23.41 \pm 1.85 \text{ b} \end{array}$	$\begin{array}{c} 21.60 \pm 0.66 \text{ a} \\ 20.63 \pm 0.46 \text{ a} \\ 19.67 \pm 0.56 \text{ a} \end{array}$	$\begin{array}{c} 62.93 \pm 1.22 \text{ a} \\ 63.65 \pm 1.20 \text{ a} \\ 63.66 \pm 1.10 \text{ a} \end{array}$	$\begin{array}{c} 9.64 \pm 0.54 \text{ a} \\ 7.14 \pm 0.50 \text{ b} \\ 6.03 \pm 0.51 \text{ b} \end{array}$
day 6, air 80% O <sub>2</sub> 100% O <sub>2</sub>	$\begin{array}{c} 23.48 \pm 1.67 \text{ a} \\ 20.04 \pm 1.15 \text{ b} \\ 15.23 \pm 1.02 \text{ c} \end{array}$	$\begin{array}{c} 19.82 \pm 0.76 \text{ a} \\ 17.17 \pm 0.46 \text{ b} \\ 14.37 \pm 0.49 \text{ c} \end{array}$	$\begin{array}{c} 55.35 \pm 1.00 \text{ a} \\ 55.76 \pm 1.20 \text{ a} \\ 58.37 \pm 1.30 \text{ a} \end{array}$	$12.21 \pm 0.52$ a 11.20 $\pm$ 0.56 a 10.08 $\pm$ 0.51 a
day 9, air 80% O <sub>2</sub> 100% O <sub>2</sub>	$\begin{array}{c} 21.07 \pm 1.72 \text{ a} \\ 17.54 \pm 1.76 \text{ a} \\ 12.56 \pm 1.70 \text{ b} \end{array}$	$\begin{array}{c} 17.67 \pm 0.66 \text{ a} \\ 15.97 \pm 0.56 \text{ b} \\ 13.21 \pm 0.51 \text{ c} \end{array}$	$\begin{array}{c} 44.02\pm 0.90 \text{ b} \\ 47.42\pm 0.74 \text{ a} \\ 46.62\pm 1.20 \text{ a} \end{array}$	$12.84 \pm 0.55$ a 11.84 $\pm$ 0.53 a 11.29 $\pm$ 0.56 a
day 12, air 80% O <sub>2</sub> 100% O <sub>2</sub>	$\begin{array}{c} \text{20.96} \pm \text{1.56 a} \\ \text{12.61} \pm \text{1.51 b} \\ \text{10.22} \pm \text{1.58 b} \end{array}$	$\begin{array}{c} 17.34 \pm 0.57 \text{ a} \\ 12.94 \pm 0.53 \text{ b} \\ 10.13 \pm 0.66 \text{ c} \end{array}$	$\begin{array}{c} 40.89 \pm 1.40 \text{ b} \\ 44.05 \pm 0.80 \text{ a} \\ 43.75 \pm 1.20 \text{ a} \end{array}$	$\begin{array}{c} \text{16.21} \pm 0.52 \text{ a} \\ \text{15.62} \pm 0.51 \text{ a} \\ \text{14.26} \pm 0.53 \text{ b} \end{array}$

<sup>*a*</sup> Data were expressed as the mean  $\pm$  SEM of triplicate assays. Values in the same column with different letters for each day were significantly different at  $P \leq 0.05$ .

perature of storage (4). The different change patterns of pH, TA, and TSS in different studies could be associated with the different effects of elevated  $O_2$  on the commodity respiratory rate.

**Respiratory Rate and Ethylene Production.** The respiratory rate of Chinese bayberry measured as CO<sub>2</sub> evolution at harvest was around 35 mg CO<sub>2</sub> kg<sup>-1</sup> h<sup>-1</sup> (**Table 2**). During the 5 °C storage period, respiration of both high O2 treated and control fruit decreased gradually. No significant differences were observed among any of the treatments in the first 3 days. After day 3, comparable respiratory rate was found among different high O<sub>2</sub> treatment groups and the air control group, and 100% O<sub>2</sub> was the most efficient in reducing respiration throughout the storage period. At the end of storage, respiratory rates of Chinese bayberry stored in air, 80% and 100%, were 21.0, 12.6, and 10.2 mg  $\text{CO}_2$  kg<sup>-1</sup> h<sup>-1</sup>, respectively. Ethylene production of all fruit was reduced from 23.7  $\mu$ L kg<sup>-1</sup> h<sup>-1</sup> initial to 13.5  $\mu$ L kg<sup>-1</sup> h<sup>-1</sup> at the end of storage on average (**Table 2**). Elevated  $O_2$  atmospheres reduced the ethylene production slightly, but significantly, from day 3 and thereafter. The improved inhibition of the respiratory rate and ethylene production in bayberries were obtained with increased O2 concentration. The lower respiration rate and ethylene production for Chinese bayberry at 100%  $O_2$  was possibly a contributing factor to its extended shelf life. Similar results were obtained in other previous research, in which 80% or 100%  $O_2$  reduced ethylene production rates, delayed ripening of mature green and breaker tomatoes at 20 °C, and also 40, 60, or 80 kPa O<sub>2</sub> inhibited the respiration rates of Bartlett pear slices and ethylene production during 4 days at 10 °C (4). These data suggest that high  $O_2$  treatments could inhibit the respiration and ethylene release of Chinese bayberry, thereby delaying the deterioration of fruit quality. This conclusion is supported by the observations that in higher O<sub>2</sub> treatments, there were higher TSS, TA content, and lower pH values of **Table 1** in the present work.

Ascorbic Acid and MDA. There was a decrease in the measurement of ascorbic acid content after harvest in all treatments, which indicates the nutrition loss in fruit with storage (**Table 2**). No significant differences in ascorbic acid loss were found between the control and high  $O_2$  treated fruits. The

Table 3. Changes in Chinese Bayberry Fruit Superoxide Dismutase (SOD), Catalase (CAT), Ascorbate Peroxidase (APX), and Peroxidase (POD) Activities during Storage at 5 °C in Air or High O<sub>2</sub> Atmosphere<sup>a</sup>

treatment	SOD activity (U mg <sup>-1</sup> protein)	CAT activity (U mg <sup>-1</sup> protein)	APX activity (U mg <sup>-1</sup> protein)	POD activity (U mg <sup>-1</sup> protein)
day 0, air	$899.36 \pm 21.27$	$44.64\pm0.96$	$14.26\pm1.30$	$80.07\pm5.00$
day 3, air 80% Oc	937.40 ± 31.01 a 960.34 ± 40.77 a	$42.77 \pm 1.06$ b $47.17 \pm 0.88$ a	$26.33 \pm 1.20$ c 30 30 $\pm$ 1 20 b	$88.54 \pm 3.14$ c 97.07 $\pm$ 2.11 b
100% O <sub>2</sub>	970.39 ± 40.85 a	49.73 ± 1.56 a	$36.54 \pm 1.00 \text{ a}$	$107.42 \pm 6.17$ a
day 6, air 80% O <sub>2</sub> 100% O <sub>2</sub>	$644.13 \pm 41.67$ a $674.52 \pm 31.15$ a $723.81 \pm 34.02$ a	$38.04 \pm 0.76$ b 40.81 $\pm$ 0.96 a 43.53 $\pm$ 1.49 a	$\begin{array}{c} 34.81 \pm 1.00 \text{ c} \\ 40.98 \pm 1.20 \text{ b} \\ 46.10 \pm 1.30 \text{ a} \end{array}$	$92.97 \pm 3.31 \text{ c}$ 107.18 $\pm$ 2.37 b 112.58 $\pm$ 2.37 a
day 9, air 80% O <sub>2</sub> 100% O <sub>2</sub>	$454.69 \pm 34.72$ b $550.63 \pm 36.76$ a $627.55 \pm 37.70$ a	$\begin{array}{c} 34.23 \pm 0.66 \text{ c} \\ 37.05 \pm 1.56 \text{ b} \\ 41.46 \pm 1.51 \text{ a} \end{array}$	$26.53 \pm 0.90$ b 34.67 $\pm$ 1.74 a 31.01 $\pm$ 1.72 a	$\begin{array}{c} 88.14 \pm 2.51 \text{ b} \\ 93.83 \pm 1.97 \text{ a} \\ 97.50 \pm 1.90 \text{ a} \end{array}$
day 12, air 80% O <sub>2</sub> 100% O <sub>2</sub>	$614.87 \pm 38.56$ b 718.66 $\pm$ 33.51 a 675.35 $\pm$ 20.58 a	$\begin{array}{c} 31.13 \pm 0.87 \text{ b} \\ 36.73 \pm 1.53 \text{ a} \\ 39.08 \pm 1.66 \text{ a} \end{array}$	$\begin{array}{c} 19.82 \pm 1.40 \text{ c} \\ 23.94 \pm 0.80 \text{ b} \\ 28.55 \pm 1.20 \text{ a} \end{array}$	$85.20 \pm 1.50$ c $89.26 \pm 1.44$ b $94.79 \pm 1.27$ a

<sup>a</sup> Data were expressed as the mean  $\pm$  SEM of triplicate assays. Values in the same column with different letters for each day were significantly different at  $P \leq 0.05$ .

changes in MDA content were proved to be similar in all treated fruits stored at 5 °C for 12 days. MDA contents of bayberry stored in both high O<sub>2</sub> atmospheres and air increased with storage time, and a marked increase in MDA content of the fruits stored in air was found (**Table 2**). High O<sub>2</sub> atmosphere storage can inhibit the accumulation of MDA with increased O<sub>2</sub> concentration. Very little information is available on the effects of elevated O<sub>2</sub> levels on concentrations of ascorbic acid and MDA in fresh-intact and fresh-cut fruits and vegetables. Day et al. (25) reported that high O<sub>2</sub> MAP had beneficial effects on the retention of ascorbic acid and the degree of lipid oxidation. In strawberries, treatments with 60% or 100% O<sub>2</sub> inhibited the loss of ascorbic acid and the accumulation of MDA (14).

Antioxidant Enzyme Activities. SOD activity increased slightly in air and high O<sub>2</sub> (80% and 100%) treated Chinese bayberry within first 3 days. The enhanced SOD activity was then decreased drastically in all treatments from day 3 to day 9 and inclined slightly at the end of the storage again. There were no clear differences between SOD activity in two high O2 treatments and control (Table 3). CAT activity of control fruit decreased 1.4-fold during 12 days storage, while for bayberries stored in high O<sub>2</sub> atmosphere, CAT activities increased within 3 days, then declined gradually with storage time. Relatively higher levels of CAT activity were found in the fruits stored in high O<sub>2</sub> atmosphere compared with those stored in air (Table **3**). APX activities increased first in both high  $O_2$  and control fruit, and then decreased gradually with a similar tendency. There were significant differences in APX activities among the air and high O<sub>2</sub> treatments during the whole storage (**Table 3**) period. POD activities of Chinese bayberry fruit in different storage atmospheres increased transiently and peaked at around day 6 before it started to decrease. High O2 atmosphere could maintain significantly higher levels of POD activity than air throughout the storage period (Table 3).

The accumulation of ROS resulting from an altered balance between ROS production and scavenging capacities will reduce the storage quality and marketability of fruits and vegetables (26). MDA is considered to be an indicator of membrane lipid peroxidation induced by oxidative stress, and SOD, CAT, APX, and POD are important ROS scavenging enzymes. High O<sub>2</sub> (70% O<sub>2</sub> concentration) treatment induced SOD and CAT activities and maintained membrane integrity in peach (27) and loquat (28) fruits. Chen et al. (14) also reported that treatments with >60% oxygen atmosphere maintained significantly higher levels of SOD, CAT, and APX activities in strawberries and inhibited the increases of superoxide radical production, MDA content, and fruit decay. Postharvested Chinese bayberry showed senescence metabolism after 3-6 days, indicated by the increase of membrane permeability and MDA content but rapid decrease of SOD activity (2). In this experiment, the membrane lipid peroxidation (MDA content) in control Chinese bayberry stored at 5 °C increased gradually, and CAT activities decreased significantly, suggesting that there might be marked oxidative stress in bayberry fruit during storage. Compared with control fruit, the degree of oxidative stress in high O<sub>2</sub> treatments might be less serious for the significantly lower MDA contents with higher CAT, APX, and POD activities during most storage periods. These data suggest that high  $O_2$  treatment may be helpful in maintaining Chinese bayberry fruit resistance to senescence development and decay incidence, which are associated with oxidative stress.

Total Phenolic, Total Anthocyanin, and DPPH Radical Scavenging Activity. The major phenolic and anthocyanin in different bayberry cultivars have been identified as gallic acid and cyanidin 3-glucoside (29). During storage, the total phenolic content in all treated fruits exhibited a slight increase during the first 3 days; thereafter, it decreased gradually during the last 9 days. There was an increase in total phenolic content with an increase in O<sub>2</sub> concentrations. No significant differences in total phenolic content were found during the first 6 days of storage in all treatments. However, there were significantly higher levels of total phenolic content in fruits treated with high O2 during the following 6 days of storage compared with that in the air control (Table 4). Total anthocyanin content increased in the first 6 days, then declined gradually during the remaining time, and showed similar change patterns in all treatments. No significant differences of total anthocyanin content were observed in all treatments during storage periods (Table 4). DPPH radical scavenging activity of Chinese bayberry showed a similar change pattern as did total phenolic and anthocyanin content during storage in response to different storage atmospheres. DPPH radical scavenging activities increased within the first 3 days and then declined gradually after they reached their peak values (Table 4). Significantly higher DPPH radical scavenging activities were obtained in high O2 treated Chinese bayberry from day 6 to the end of storage than in air.

 Table 4. Changes in Chinese Bayberry Fruit Total Phenolic, Total

 Anthocyanin and DPPH-Radical Scavenging Activity during Storage at 5

 °C in Air or High O<sub>2</sub> Atmosphere<sup>a</sup>

treatment	total phenolic <sup>b</sup> (mg g <sup>-1</sup> FW)	total anthocyanin <sup>c</sup> (mg g <sup>-1</sup> FW)	DPPH-radical scavenging activity <sup>d</sup> (mg g <sup>-1</sup> FW)
day 0, air	$8.27\pm0.15$	$4.20\pm0.17$	$180.5\pm7.4$
day 3, air	$8.55 \pm 0.75$ a	$4.27 \pm 0.15  a$	$183.3 \pm 2.8 \ { m a}$
80% O <sub>2</sub>	$8.93\pm0.45~\mathrm{a}$	$4.37\pm0.13$ a	$186.2 \pm 2.6 \ { m a}$
100% O <sub>2</sub>	$9.75 \pm 0.57 \ { m a}$	$4.36\pm0.14$ a	$192.3 \pm 4.1 \ { m a}$
day 6, air	$7.39\pm0.11$ b	$4.25\pm0.14~\mathrm{a}$	$168.7\pm5.3$ b
80% O <sub>2</sub>	$7.50\pm0.11$ b	$4.32 \pm 0.15  a$	$176.8\pm4.6$ ab
100% O <sub>2</sub>	$8.06\pm0.23~\mathrm{a}$	$4.36 \pm 0.11  a$	$180.0 \pm 4.4 \ { m a}$
day 9, air	$6.12\pm0.14~\mathrm{c}$	$4.10\pm0.12$ a	$155.0\pm4.1$ b
80% O <sub>2</sub>	$6.93\pm0.24$ b	$4.14 \pm 0.11  \mathrm{a}$	$172.5 \pm 2.9 \ { m a}$
100% O <sub>2</sub>	$7.58 \pm 0.34$ a	$4.23 \pm 0.14 \ { m a}$	$177.3 \pm 3.2 \text{ a}$
day 12, air	$5.06\pm0.14~\mathrm{c}$	$3.77\pm0.14$ b	$115.5\pm6.5~\mathrm{c}$
80% O <sub>2</sub>	$6.03\pm0.11$ b	$3.90\pm0.11$ ab	$149.0\pm2.5$ b
100% O <sub>2</sub>	$\textbf{6.98} \pm \textbf{0.17}~\textbf{a}$	$4.07\pm0.16~\text{a}$	$173.9 \pm 8.1 \ { m a}$

<sup>*a*</sup> Data were expressed as mean  $\pm$  SEM of triplicate assays. <sup>*b*</sup> Data expressed as milligrams of gallic acid equivalents per gram of fresh weight. <sup>*c*</sup> Data expressed as milligrams of cyanidin 3-glucoside equivalents per gram of fresh weight. <sup>*d*</sup> Data expressed as milligrams of  $\alpha$ -tocopherol equivalents per gram of fresh weight. Values in the same column with different letters for each day were significantly different at  $P \leq 0.05$ .

In blueberries, the antioxidant activity was markedly increased by 60% - 100% oxygen treatments as compared with 40% O<sub>2</sub> treatment and air control during 35 days of storage at 5 °C. Meanwhile, the elevated O2 between 60% and 100% also promoted increases of total phenolic and total anthocyanin content (7).  $O_2$  concentrations higher than 60 kPa promoted increases in oxygen radical absorbance capacity values, total phenolics, and anthocyanins in strawberry during the initial 7 days of storage at 5 °C, but this effect diminished with prolonged storage (30); Pérez and Sanz (24) found that, in comparison with fruits stored in air, strawberries held in 80%  $O_2 + 20\%$ CO<sub>2</sub> had significantly higher levels of total anthocyanin during the first 4 days but significantly lower levels of total anthocyanin at the end of storage. These results suggest that the effect of high O2 atmosphere on total phenolics, total anthocyanins, and antioxidant activities in berry crops may vary depending on the commodity, O<sub>2</sub> concentration, storage duration, and temperature. In present study, Chinese bayberry stored in high O<sub>2</sub> atmosphere (80% and 100%) had little effect on total anthocyanin content. However, bayberry fruit stored in 80% and 100% O2 atmosphere exhibited significantly higher levels of total phenolic content and DPPH radical scavenging activity compared to those in air treated fruit. Our results indicate that storage atmosphere enriched with O<sub>2</sub> from 80% to 100% will improve the health benefit and antioxidant status of Chinese bayberry by positively affecting phenolic metabolism to improve DPPH radical scavenging activity.

The assay of DPPH radical scavenging activity and total phenolic and anthocyanin levels measures the overall antioxidant capacity of Chinese bayberry against ROS. Enzymatic activities of ROS scavenging were enhanced by elevated  $O_2$  level during the storage period. Interestingly, the antioxidative capacity of Chinese bayberry was also enhanced as indicated by the DPPH radical scavenging activity and total phenolic and anthocyanin level measurements. All these results indicate that the elevated  $O_2$  level increased the ability of the antioxidative defense mechanism in Chinese bayberry in order to first control the ROS level and eventually the fruit decay severity. However, this situation was deficient after 3 days of storage when the ROS scavenging enzymes activity and fruit overall antioxidant

capacity decreased and continued throughout the remainder of the storage period. These drops possibly led to an acceleration of senescence development and an increase in fruit decay incidence. Storage with 80% and 100%  $O_2$  atmosphere seems to be just as effective in enhancing enzyme activity, total phenolic and anthocyanin levels, and DPPH radical scavenging activity in bayberry fruit.

In summary, fruit decay and quality deterioration of Chinese bayberry were reduced by both 80% and 100%  $O_2$  treatments. This reduction was promoted with increases in  $O_2$  concentrations. Chinese bayberry fruit treated with 100%  $O_2$  consistently maintained the lowest fruit decay index and the highest antioxidative enzyme activities, DPPH radical scavenging activity, and total phenolic and anthocyanin contents.

# LITERATURE CITED

- Xi, Y. F.; Zheng, Y. H.; Qian, D. M; Ying, T. J. Effects of storage temperature on changes in nutritional composition and decay rates in fruit of red bayberry. *Bull. Sci. Technol.* **1993**, *9*, 254–256, In Chinese with English abstract.
- (2) Xi, Y. F.; Luo, Z. S.; Xu, C.; Chen, D.; Wang, Y. G. Effects of CA storage on active oxygen metabolism in Chinese bayberry fruit (*Myrica rubra*). J. Zhejiang Univ. (Agric. <u>Life Sci.</u>) 2001, 27, 311–313, In Chinese with English abstract.
- (3) Qi, X. J.; Wang, L. P.; Liang, S. M.; Zhou, L. Q.; Cai, X. Q. Study on the mycoflora in bayberry (*Myrica rubra*) fruits treated by controlled atmosphere storage. *Acta Agric. Zhejianggensi.* 2003, 15, 28–30, In Chinese with English abstract.
- (4) Kader, A. A.; Ben-Yehoshua, S. Effects of superatmospheric oxygen levels on postharvest physiology and quality of fresh fruits and vegetables. *Postharvest Biol. Technol.* 2000, 20, 1–13.
- (5) Van der Steen, C.; Jacxsens, L.; Devlieghere, F.; Debevere, J. Combining high oxygen atmospheres with low oxygen modified atmosphere packaging to improve the keeping quality of strawberries and raspberries. *Postharvest Biol. Technol.* 2002, *26*, 49–58.
- (6) Wszelaki, A. L.; Mitcham, E. J. Effects of superatmospheric oxygen on strawberry fruit quality and decay. <u>Postharvest Biol.</u> <u>Technol.</u> 2000, 20, 125–133.
- (7) Zheng, Y. H.; Wang, C. Y.; Wang, S. Y.; Zheng, W. Effect of high-oxygen atmospheres on blueberry phenolics, anthocyanins, and antioxidant capacity. *J. Agric. Food Chem.* **2003**, *51*, 7162– 7169.
- (8) Amanatidou, A.; Smid, E. J.; Gorris, L. G. M. Effect of elevated oxygen and carbon dioxide on the surface growth of vegetableassociated microorganisms. <u>J. Appl. Microbiol</u>, 1999, 86, 429– 438.
- (9) Fridovich, I. Biological effects of the superoxide radical. <u>Arch.</u> <u>Biochem. Biophys.</u> 1986, 247, 1–11.
- (10) Møller, I. M. Plant mitochondria and oxidative stress: electron transport, NADPH turnover, and metabolism of reactive oxygen species. <u>Annu. Rev. Plant Physiol. Plant Mol. Biol</u>. 2001, 52, 561– 591.
- (11) Mittler, R. Oxidative stress, antioxidants and stress tolerance. *Trends Plant Sci.* **2002**, *7*, 405–410.
- (12) Karpinski, S.; Wingsle, G.; Karpinska, B.; Hallgren, J. Molecular responses to photooxidative stress in *Pinus sylvestris* (L.). II. Differential expression of CuZn-superoxide dismutases and glutathione reductase. *Plant Physiol.* **1993**, *103*, 1385–1391.
- (13) Dipierro, S.; Leonardis, S. D. The ascorbate system and lipid peroxidation in stored potato (*Solanum tuberosum* L.) tubers. <u>J.</u> <u>Exp. Biol.</u> 1997, 48, 779–783.
- (14) Chen, X. H.; Zheng, Y. H.; Yang, Z. F.; Ma, S. J.; Feng, L.; Wang, X. X. Effects of high oxygen treatments on active oxygen metabolism and fruit decay in postharvest strawberry. *J. Naniing Agric. Univ.* 2005, 28, 99–102, In Chinese with English abstract.
- (15) Jones, E.; Hughes, R. E. Foliar ascorbic acid in some angiosperms. <u>*Phytochem.*</u> 1983, 22, 2493–2499.

- (16) Bradford, M. N. A rapid and sensitive method for the quantitation of microgram quantities of protein utilizing the principle of protein-dye binding. *Anal. Biochem.* **1976**, 72, 248–254.
- (17) Rao, M. V.; Paliyath, G.; Ormrod, D. P. Ultraviolet-B- and ozoneinduced biochemical changes in antioxidant enzymes of *Arabidopsis thaliana*. *Plant Physiol*. **1996**, *110*, 125–136.
- (18) Beers, R. F.; Sizer, I. W. A spectrophotometric method for measuring the breakdown of hydrogen peroxide by catalase. *J. Biol. Chem.* **1952**, *195*, 133–140.
- (19) Nakano, Y.; Asada, K. Hydrogen peroxide is scavenged by ascrobate specific peroxidase in spinach chloroplasts. *Plant Cell Physiol.* **1989**, 22, 867–880.
- (20) Kochba, J.; Lavee, S.; Spiege, R. P. Difference in peroxidase activity and isoenzymes in embryogenic and non-embryogenic 'Shamouti' orange ovular callus lines. *Plant Cell Physiol.* **1977**, *18*, 463–467.
- (21) Cheng, G. W.; Breen, P. J. Activity of phenylalanine ammonialyase (PAL) and concentration of anthocyanins and phenolics in developing strawberry fruit. <u>J. Amer. Soc. Hortic. Sci</u>. **1991**, *116*, 865–869.
- (22) Slinkard, K.; Singleton, V. L. Total phenol analysis: automation and comparison with manual methods. <u>Am. J. Enol. Vitic</u>. 1977, 28, 49–55.
- (23) Larrauri, J. A.; Sanchez-Moreno, C.; Saura-Calixto, F. Effect of temperature on the free radical scavenging capacity of extracts from red and white grape pomace peels. <u>J. Agric. Food Chem.</u> 1998, 46, 2694–2697.
- (24) Pérez, A. G.; Sanz, C. Effect of high-oxygen and high-carbondioxide atmospheres on strawberry flavor and other quality traits. *J. Agric. Food Chem.* 2001, 49, 2370–2375.

- (25) Day, B. P. F.; Bankier, W. J.; Gonzalez, M. I. Novel Modified Atmosphere Packaging (MAP) for Fresh Prepared Produce. Research Summary, Sheet 13; Campden and Chorleywood Food Research Association: Chipping Campden, U.K., 1998.
- (26) Hodges, D. M.; Lester, G. E.; Munro, K. D.; Toivonen, P. T. A. Oxidative stress: importance for postharvest quality. *HortScience* 2004, *39*, 924–929.
- (27) Wang, Y. S.; Tian, S. P.; Xu, Y. Effects of high oxygen concentration on pro- and anti-oxidant enzymes in peach fruits during postharvest periods. *Food Chem.* 2005, *91*, 99–104.
- (28) Ding, Z. S.; Tian, S. P.; Wang, Y. S.; Li, B. Q.; Chan, Z. L.; Han, J.; Xu, Y. Physiological response of loquat fruit to different storage conditions and its storability. *Postharvest Biol. Technol.* 2006, *41*, 143–150.
- (29) Fang, Z. X.; Zhang, M.; Wang, L. X. HPLC-DAD-ESIMS analysis of phenolic compounds in bayberries (*Myrica rubra* Sieb. et Zucc.). *Food Chem.* 2007, 100, 845–857.
- (30) Zheng, Y. H.; Wang, S. Y.; Wang, C. Y.; Zheng, W. Changes in strawberry phenolics, anthocyanins, and antioxidant capacity in response to high oxygen treatments. <u>*LWT-Food Sci. Technol.*</u> 2007, 40, 49–57.

Received for review September 26, 2008. Revised manuscript received November 17, 2008. Accepted November 17, 2008. This study was supported by the International Foundation for Science (E/3942-1), National Natural Science Foundation of China (No. 30170661), and Ministry of Science and Technology of China (2006BAD30B03).

JF803007J